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1.11 The force on a spring is described by:  $F = K x$  where  $K$  is the spring constant. First calculate  $K$  based on the earth measurement then  $g$  based on spring measurement on Mars.

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Chemical Engineering Thermodynamics  
 $K_{wact} = 125 \text{ mol}$  The solution is  $T_f = 549.39 \text{ K}$   $c_j \text{ mol} \square \text{ K}$  solution The actual is work 499.14K. is 25% Then greater h (b) Repeat the calculation with a temperature-dependent heat capacity  $\square 10-2 \text{ T} - 3499 \square 10-5 \text{ T}^2 + 7.464 \square 10-9 \text{ T}^3 \dots$   $CP(T) = 22.243 + 5977$  Assuming reversibility  $T_f = 479.44 \text{ K}$ .

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