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Formulas 1. The percent composition of a compound is the percent by mass of each of the elements in a compound. 2.

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(160 flashcards) The sum of the atomic masses of all the  
The number of atoms in

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one mole of an e<sup>-</sup> Which compound has a molar mass of 174.

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expressed in grams per mole (g/mol) Gram-formula mass. The mass of a mole of formula units in a ionic compound bond. Mole. the simplest structural unit of an element or compound.

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STUDY. PLAY. Mole. SI base unit used to measure the amount of a substance and communicates how much of a substance is present. Avogadro's number.  $6.022 \times 10^{23}$

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molecules. Molar mass. Mass in grams of one mole of any pure substance, has the units g/mol.

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uranium-235 can ...

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material from the primary book and enhance mastery of this complex subject. Organic chemistry is a constantly evolving field that has great relevance for all scientists, not just chemists. For chemical engineers, understanding the properties of organic molecules and how reactions occur is critically important to

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understanding the processes in an industrial plant. For biologists and health professionals, it is essential because nearly all of biochemistry springs from organic chemistry. Additionally, all scientists can benefit from improved critical thinking and problem-solving skills that are developed from the study of organic

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chemistry. Organic chemistry, like any "skill", is best learned by doing. It is difficult to learn by rote memorization, and true understanding comes only from concentrated reading, and working as many problems as possible. In fact, problem sets are the best way to ensure that concepts are not only well understood,



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but can also be applied to real-world problems in the work place. Helps readers learn to categorize, analyze, and solve organic chemistry problems at all levels of difficulty Hundreds of fully-worked practice problems, all with solutions Key concept summaries for every chapter reinforces core content from the

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companion book

This supplement includes, for each chapter, a brief overview, activities and practice problems to reinforce skills, and a practice test. The answers section includes

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answers for all odd-numbered end-of-chapter exercises.

While other guides only have around 200 questions, this guide provides 800 questions so that the student is fully prepared before writing the exam. This book is also different from others because

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it includes full explanations along with answers to questions. Questions are included in the relevant sections mentioned below. AFOQT Exam Prep combines theory with test questions so that students can score high on the Air Force Officer Qualifying Test. The strategies included in this complete guide will help

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This third edition continues to innovate by providing students with an integrated and modern approach to the subject. The text

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emphasizes the modern tools of chemistry while incorporating historical evidence, and its unique molecular/quantitative emphasis is further reinforced by an integrated media package developed by the authors. Also of benefit is the just-in-time presentation of key content - only providing details once they are needed.



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While key topics and analytical techniques have been updated, there is now an additional, third chapter on chemical equilibrium. The authors have also developed an expanded and more integrated problem-solving emphasis that now incorporates a 4-step strategy throughout, complete with text icons. The

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whole is backed by a range of supplements, including a new illustration program, a tutorial CD, interactive learningware, an extensive Web CT component, an instructor's resource CD, and a solution CD.

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This General, Organic and Biochemistry text has been written for students preparing for careers in health-related fields such as nursing, dental hygiene, nutrition, medical technology and occupational therapy. It is also suited for students majoring in other fields where it

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is important to have an understanding of the basics of chemistry. Students need have no previous background in chemistry, but should possess basic math skills. Raymond was crafted to take advantage of recent trends in the GOB market. It is a shorter, lighter book with a new, integrated table of contents that

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develops general, organic, and biochemistry topics together, rather than in isolation. In introducing GOB material, this text uses an integrated approach in which related general chemistry, organic chemistry, and biochemistry topics are presented in adjacent chapters. This approach helps students see the strong

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connections that exist between these three branches of chemistry, and allows instructors to discuss these, interrelationships while the material is still fresh in students' minds. This integration involves the following sets of chapters: \* Chapter 3 (Compounds) and Chapter 4 (An Introduction to Organic Compounds).

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An introduction to bonding and compounds is followed by a look at the members of a few key organic families. \* Chapters 3, 4 and 6.(Reactions).. A study of inorganic.and organic compounds is followed (after a look at gases, liquids, and solids in Chapter 5) by an introduction to their reactions. \* Chapter 7 (Solutions) and

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Chapter 8 (Lipids and Membranes) A discussion of solubility is followed by a look at the importance of solubility in biochemistry. Some reactions from Chapter 6 are reintroduced. \* Chapter 9.(Acids and Bases) and Chapter 10 (Carboxylic Acids, Phenols and Amines) Principles of acid/base Chemistry from an



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inorganic perspective are followed by a chapter on the organic and biochemical aspects of this topic. \* Chapter 11 (Alcohols, Aldehydes and Ketones) and Chapter 12 (Carbohydrates). An introduction to the chemistry of alcohols, aldehydes and ketones is followed by a presentation of related biochemical

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applications.

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problem solving skills. Many chapters in this edition has been rewritten and rearranged to vitalize the topics and to include interesting examples, analogies, and images.

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