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charge. Information A half-life ($t_{1/2}$) is the time required for one-half of the nuclei of a radioisotope sample to decay to products. After each half-life, half of the existing radioactive atoms (parent element) have decayed into atoms of a new element (daughter element). Equation [] =

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isotopes to decay is called the half-
life ($t_{1/2}$). Critical Thinking Questions:

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This page is an activity meant to be worked through as a small group of ideally 4 people. Introduction to nuclear chemistry with a discussion of

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the four fundamental forces of physics (gravity, electro-magnetism, strong nuclear force, and weak nuclear force. The text covers the forces at work in an atomic nucleus and how they can lead to a balanced or unbalanced situation.

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The half-life of strontium-90 is 25

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years. How much strontium-90 will remain after 100 years if the initial amount is 4.0 g? If the mass of a radioactive substance is 8 grams and it has half life of 4 hours, how much mass remains after 8 hours?

Phosphorus-32 is radioactive and has a half life of 14.3 days.

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